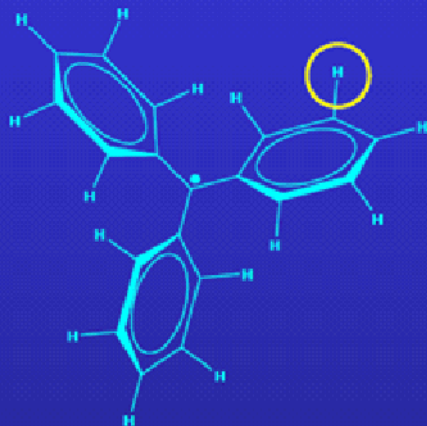
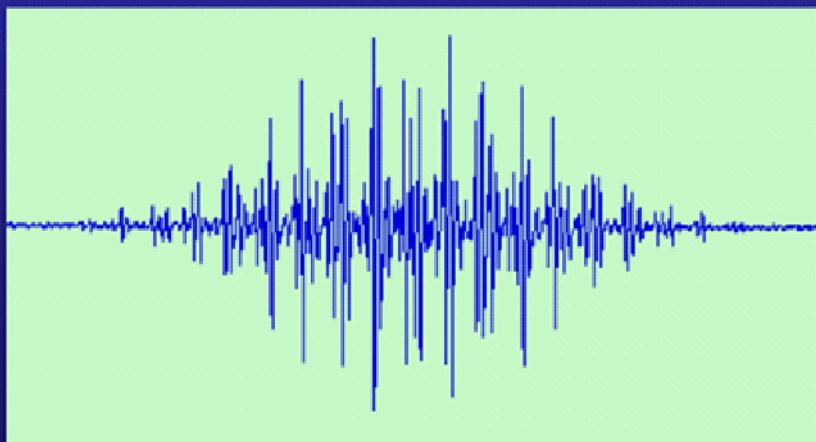


Triaryl Radikale



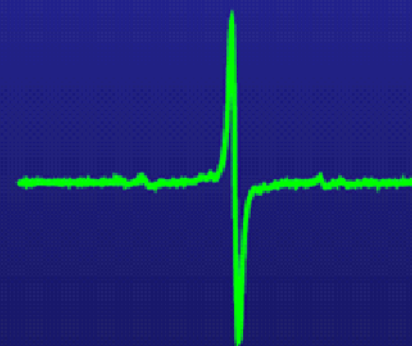
Triphenylmethyl radical



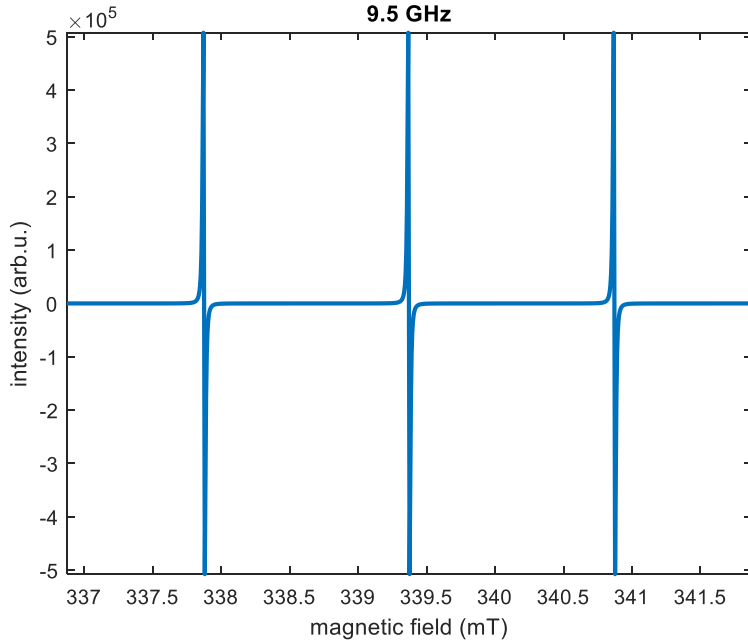
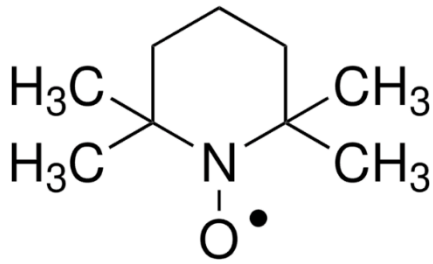
Gomberg, M. J. Am. Chem. Soc. 1900, 22, 757.



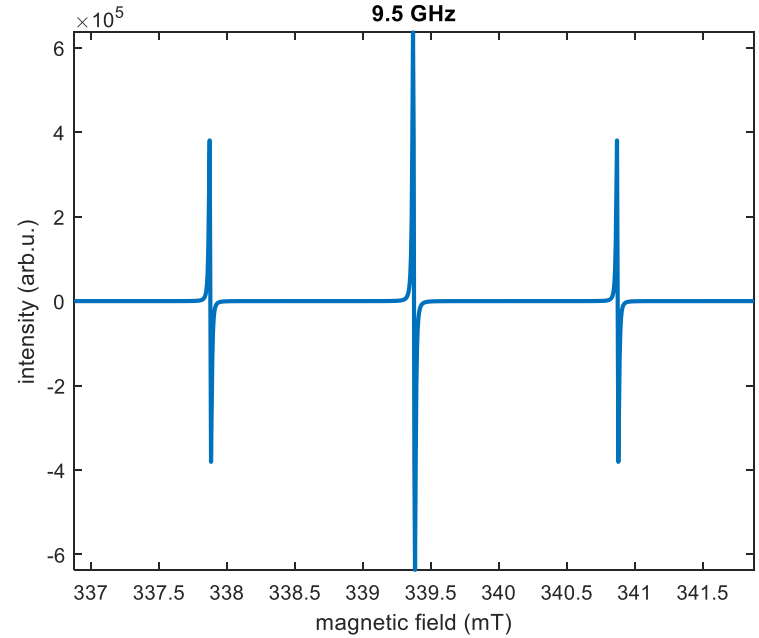
tetrathiatriarylmethyl radical (TAM)



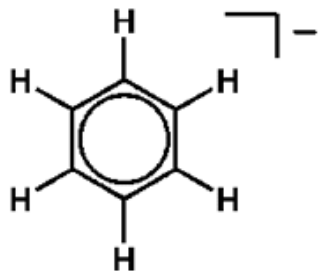
1990's Nycomed



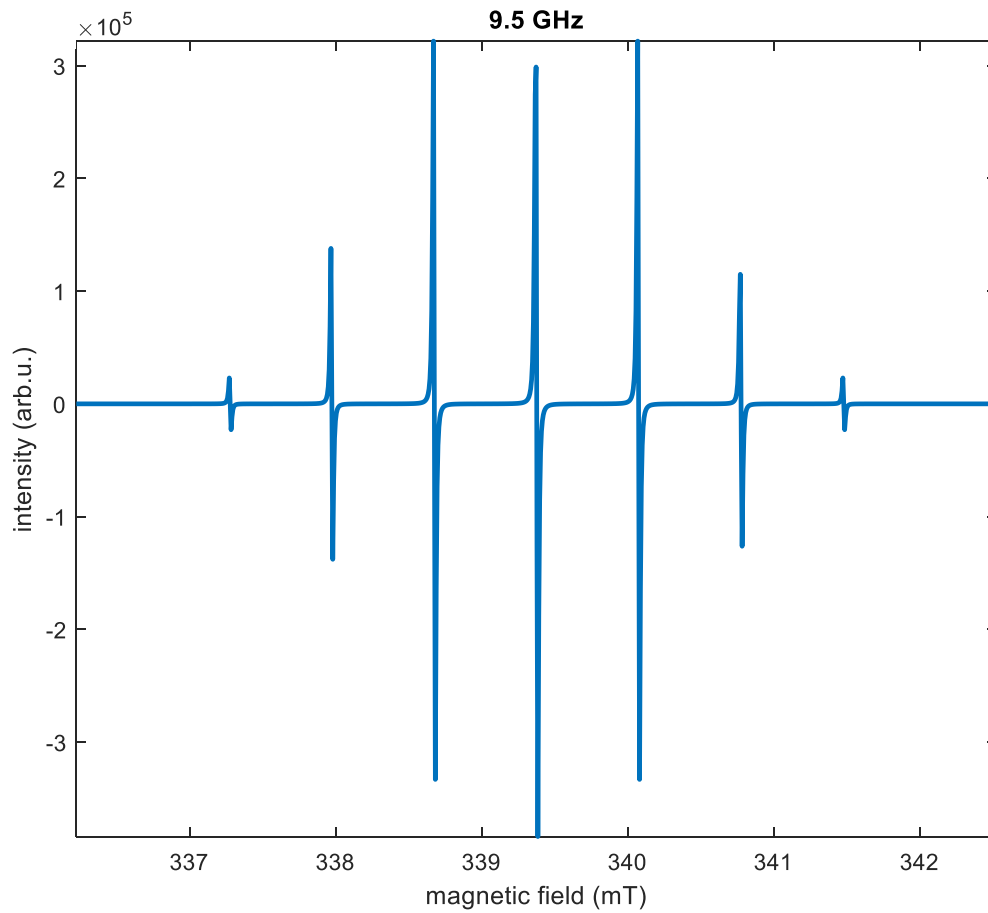
HF-Kopplung zu
1 ¹⁴N Stickstoff Kern (I=1)



HF-Kopplung zu
2 äquivalente Protonen (I=1/2)

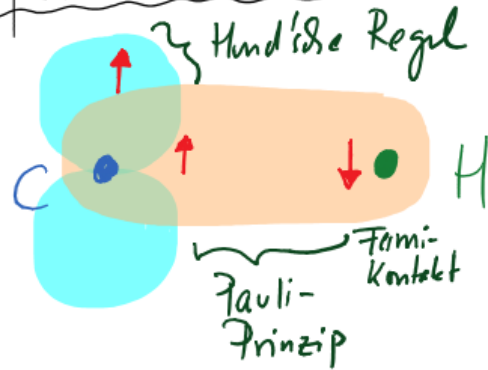


Benzol Anion Radikal



Spinpolarisation

Spinpolarisation bei π -Radikalen



Der ungepaarte e^- -Spin im π -Orbital ist bevorzugt parallel zu dem nahen e^- -Spin im σ -Orbital

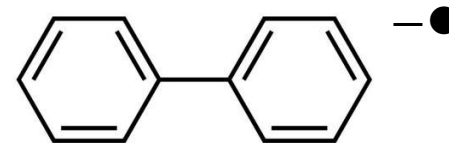
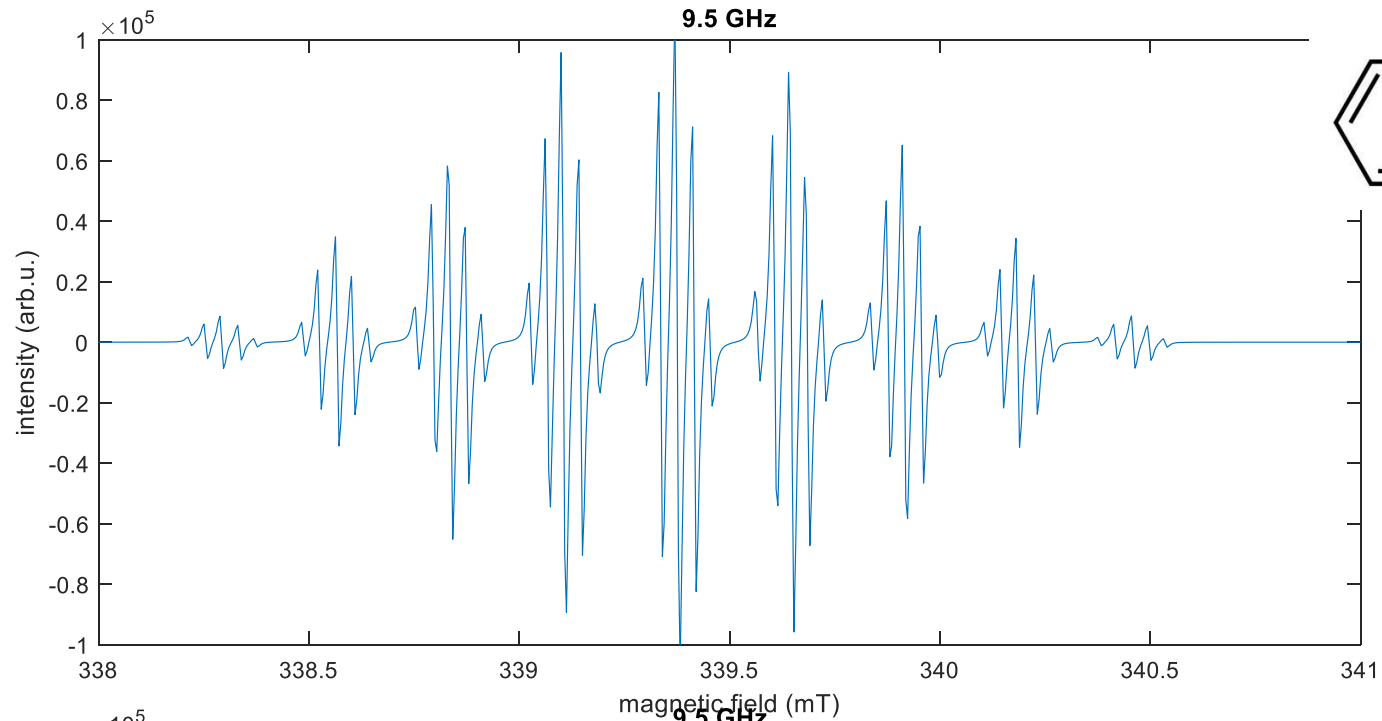
$$\hookrightarrow a_{iso}^H \neq 0 \quad \left\{ \text{obwohl } \psi_{\pi}(H) = 0 \right\}$$

McConnell Beziehung

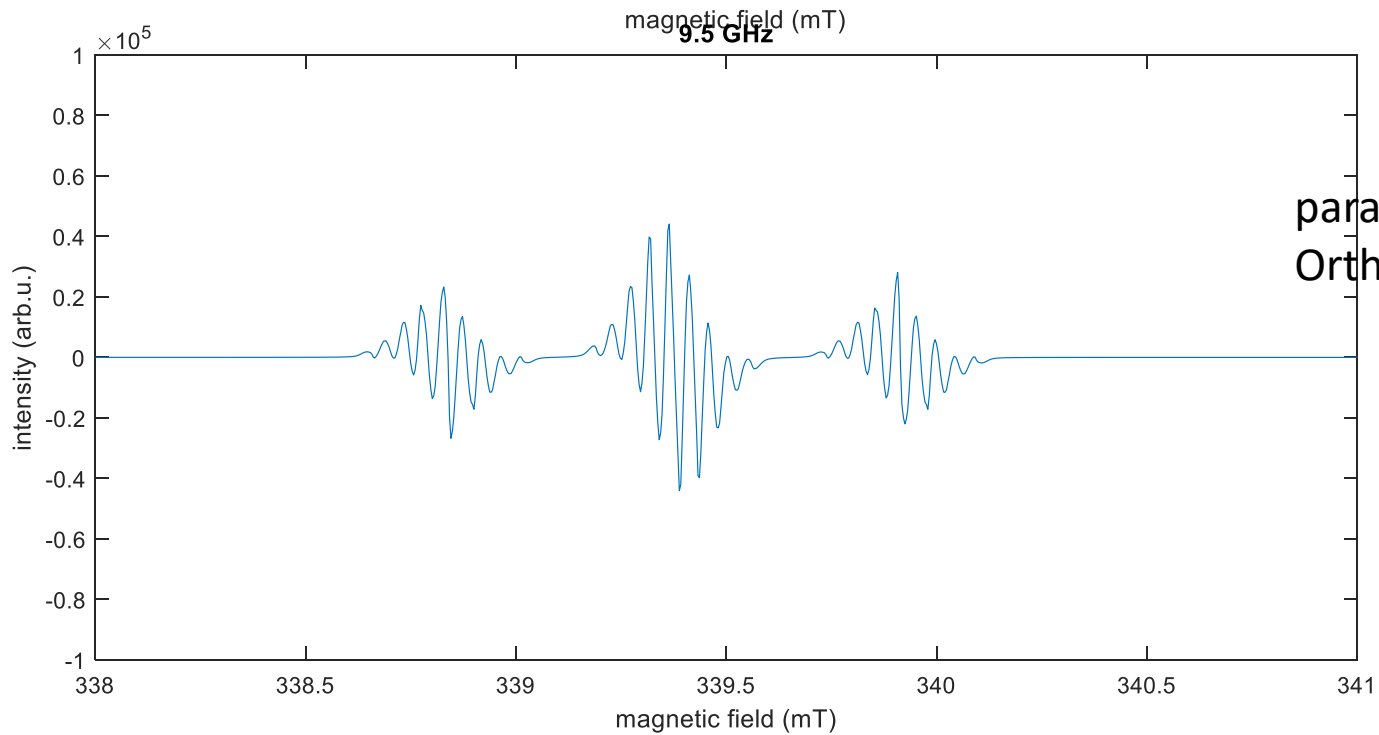
$$a_{iso}(H) = Q \cdot \underbrace{p(C)}_{\text{ungepaarte Elektronenspin-Dichte}}$$

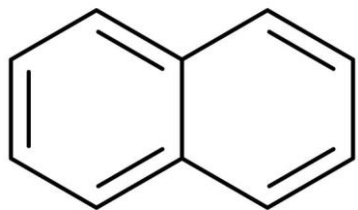
$$Q: \text{McConnell-Faktor} \approx 2.3 \text{ mT}$$

$p(C)$ z. Bsp. aus Hückel MO-Theorie oder direkt aus QC-Berechnungen (Gaussian, ORCA)

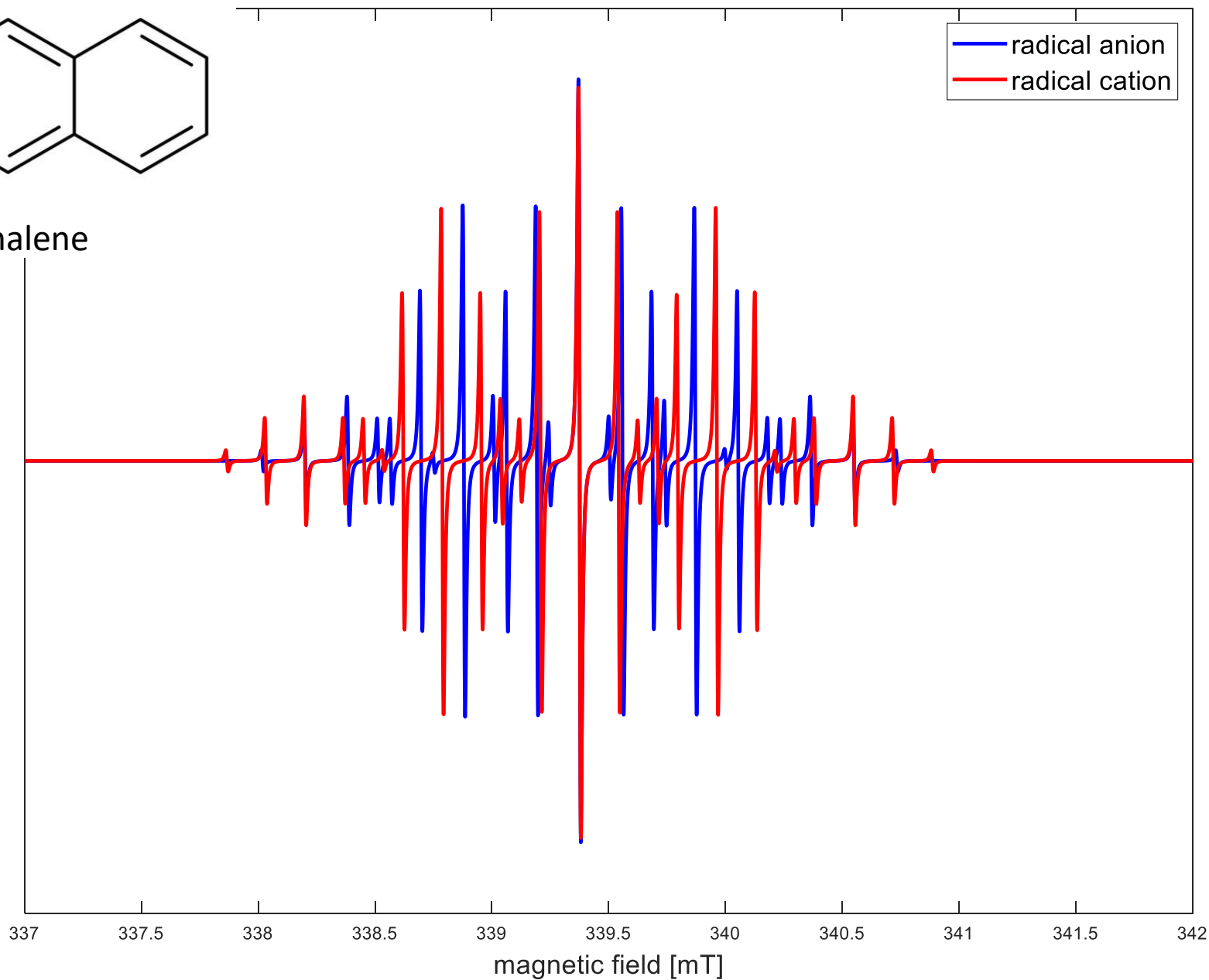


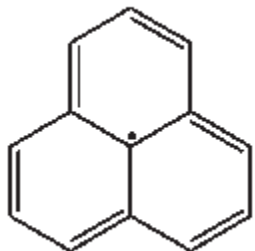
Biphphenyl Anion Radical



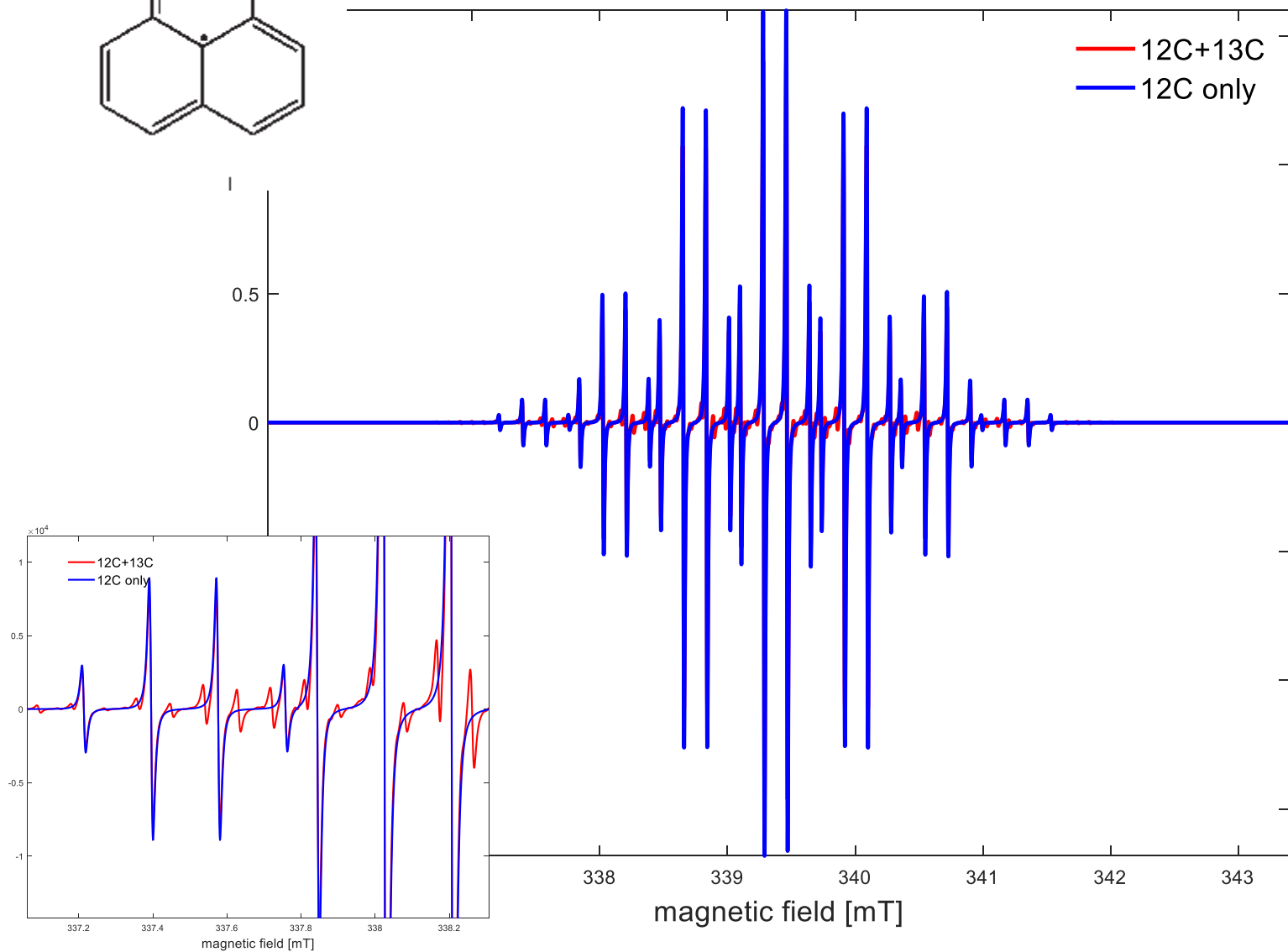


Napthalene



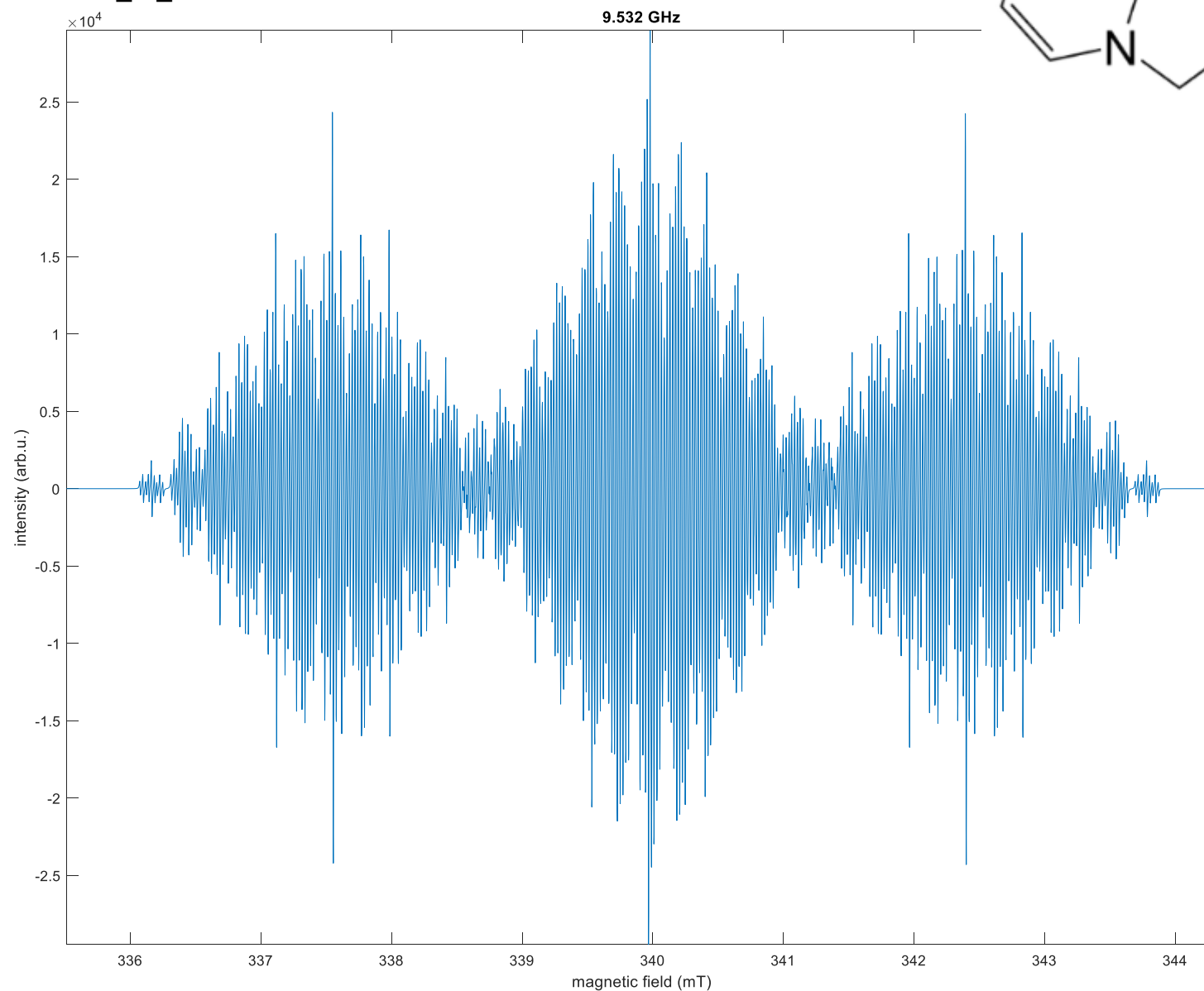
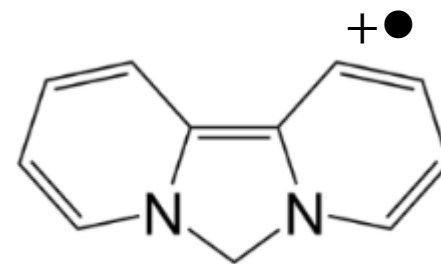


Phenalenyl Radical



Biaryl cation radical (6-hydrodipyrido[1,2-c:2',1'-e]-imidazole)

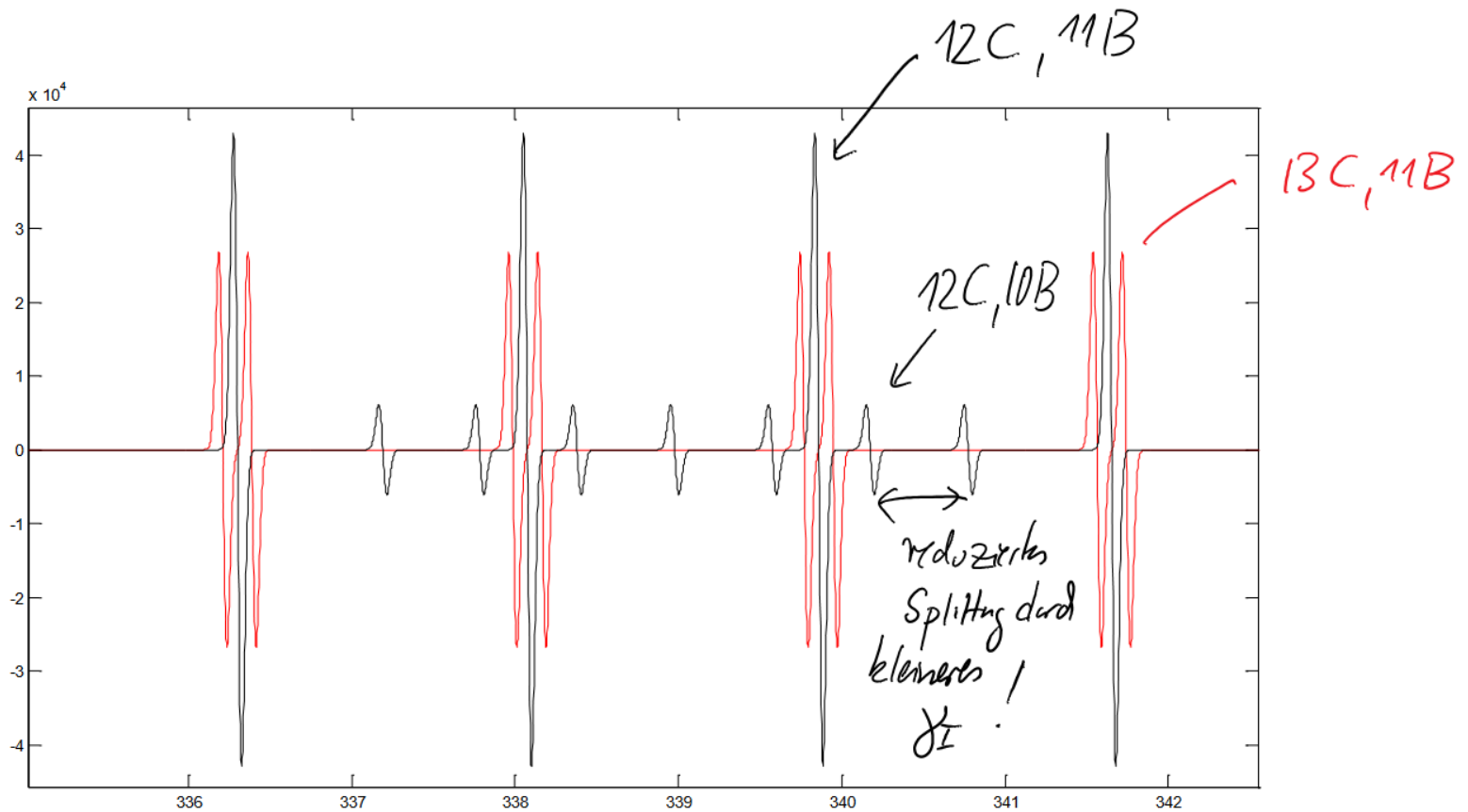
$$N = \prod (2n_j \cdot I_j + 1) = 5 \cdot 3 \cdot 3 \cdot 3 \cdot 3 \cdot 3 = 972$$



Isotopen-Signaturen

Beispiel: **B** 19.9% ^{10}B ($I=3, \frac{\gamma_I}{\gamma_S} = -1.6 \cdot 10^{-4}$) 80.1% ^{11}B ($I=3/2, \frac{\gamma_I}{\gamma_S} = -4.9 \cdot 10^{-4}$)

C 1.07% ^{13}C ($I=1/2, \frac{\gamma_I}{\gamma_S} = -3.8 \cdot 10^{-4}$)



Easy-Spin

Free Simulation Software for EPR

[EasySpin - EPR spectrum simulation](#)

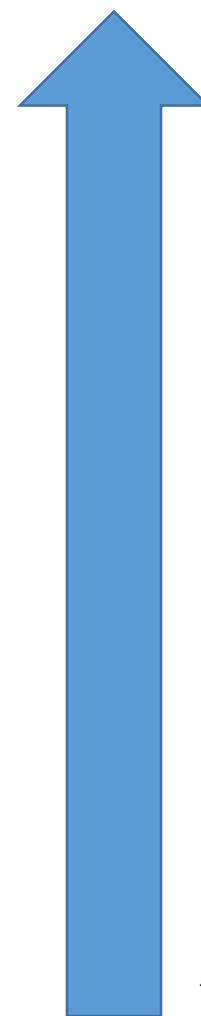
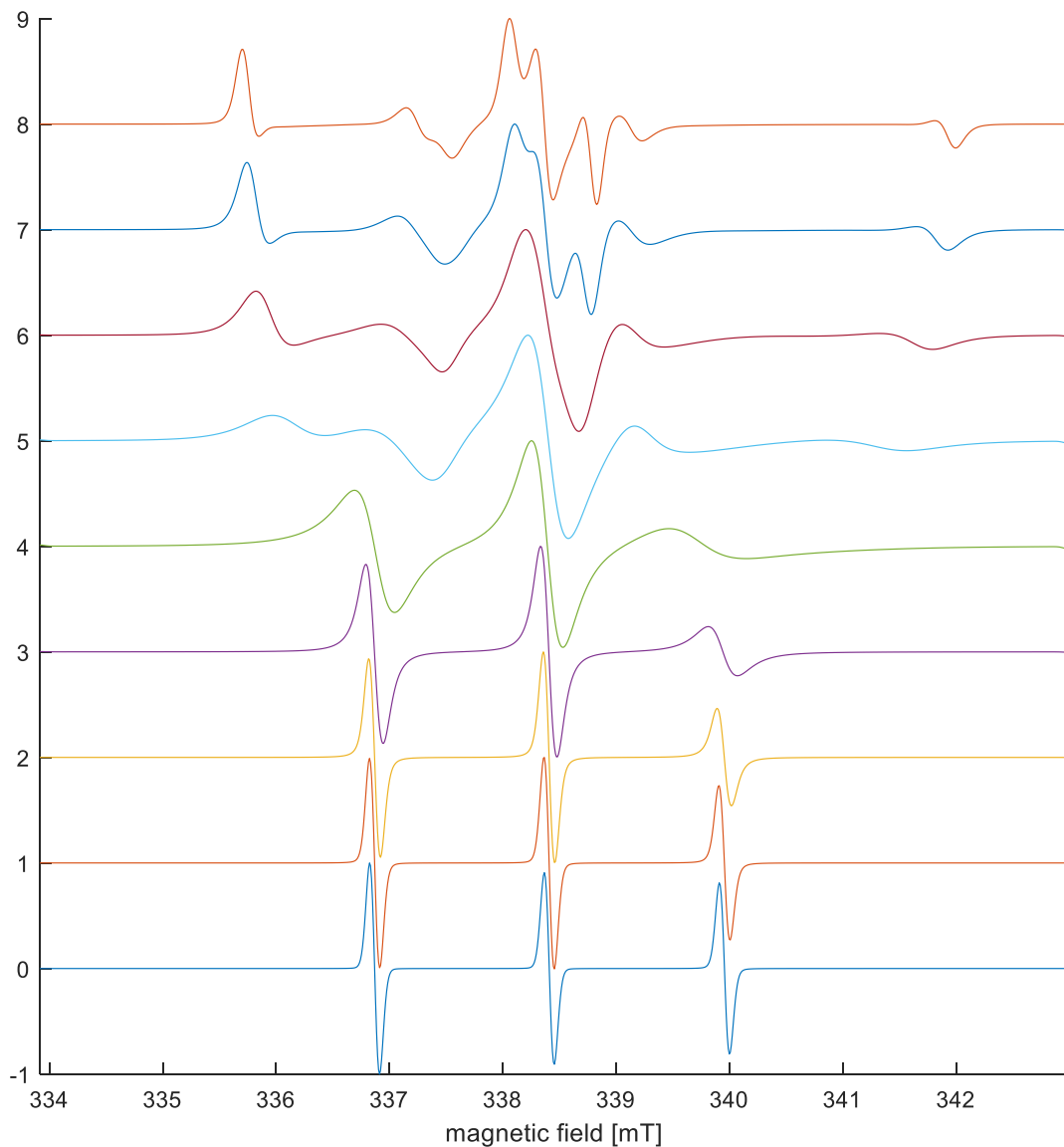
<https://easyspin.org>

Auf Matlab basierend



Stefan Stoll
University of Washington
Seattle, USA

Nitroxide Radikal bei unterschiedlich viskosen Lösemitteln



$\tau_R = 300 \text{ ns}$

$$\tau_R = \frac{V \eta}{k_B T}$$

$\tau_R = 30 \text{ ps}$

Nitroxide Radikal bei unterschiedlich viskosen Lösemitteln

